Assignments of Several Groups of Iodine (I_2) Lines in the B-X System

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The spectra obtained by means of Fourier spectroscopy and the assignments of the B-X lines of I₂ in the vicinity of two argon ion laser lines (5145 and 5287 Å), three krypton ion laser lines (5208, 5308, and 5683 Å), and one He-Ne laser line (6119 Å) are given. A detailed comparison, in the vicinity of the argon ion laser line (5145 Å), between the iodine wavenumbers calculated by means of the two sets of molecular constants previously published [Wei and Tellinghuisen, J. Mol. Spectrosc. 50, 317-332 (1974); Barrow and Yee, J. C. S. Faraday II.69, 684-700 (1973)] with those calculated from Fourier spectroscopy data, is presented.

1. INTRODUCTION

Since the recording of the iodine absorption spectrum by means of Fourier spectroscopy, and its publication as an Atlas (1), there was much interest expressed in the assignments of small portions of spectra located near some Ar⁺, Kr⁺, and He-Ne laser lines. Indeed, these laser radiations are often used to produce excited resonance fluorescence of the visible system of iodine ($B^{3}\Pi_{0t} - X^{1}\Sigma_{g}^{+}$), or to resolve the hyperfine structure of iodine lines and, more recently, to induce stimulated emission in the optically pumped iodine molecule itself (2-4). We therefore present here detailed assignments of small portions of the iodine absorption spectrum in the vicinity of two argon ion laser lines $\lambda = 5145$ Å (19 429 cm⁻¹), $\lambda = 5287$ Å (18 908 cm⁻¹), three krypton ion laser lines, $\lambda = 5208$ Å (19 192 cm⁻¹), $\lambda = 5308$ cm⁻¹ (18 830 cm⁻¹), $\lambda = 5683$ Å (17 594 cm⁻¹), and one He-Ne laser line $\lambda = 6119$ Å (16 340 cm⁻¹). These identified small portions of the iodine spectrum are also given here to show the progress achieved in the assignments of the iodine lines, and to serve as an illustration of the accuracy of the molecular constants describing the entire B-X system of I₂ (5).

2. PROCEDURE USED FOR THE LINE ASSIGNMENTS

A detailed description of the rotational analysis of a typical band (the 30, 0 band) was published in 1977 (6). Starting from previously available constants (7-10), we begin to identify lines belonging to a given band. Of course, discrepancies between our observed wavenumbers and those calculated by means of these previous constants always occur. Fortunately, these discrepancies—nearly constant at the head of the bands—increase continuously, with increasing J. The identification of the lines is also based on the intensities of the lines and the "doublet" separation as explained in Ref. (6). When a sufficient number of iden-





tified lines is obtained, we then recalculate the molecular constants describing the band under investigation by a least-squares procedure (nonlinear SIMPLEX). Using these new molecular constants we extend the identification process to higher J transitions; in general, three iterations were enough to identify lines not buried in



FIG. 2. Assignments of the iodine absorption lines in the vicinity of the Ar⁺ laser line $\lambda = 5287$ Å ($\sigma = 18$ 908 cm⁻¹).

the noise present in our spectrogram (lines with a signal to noise greater than 4). In this way we identified nearly the entire iodine absorption spectrum observed under our experimental conditions (1). Figures 1 and 2 show the iodine spectrum in the vicinity of the two argon ion laser lines and Tables I and II give the observed

312



FIG. 3. Assignments of the iodine absorption lines in the vicinity of the Kr⁺ laser line $\lambda = 5208$ Å ($\sigma = 19$ 192 cm⁻¹).

and calculated wavenumbers. Figures 3, 4, 5, and Tables III, IV, and V present the same data for the three krypton lines, and Fig. 6 and Table IV for the He-Ne laser line. The calculated wavenumbers given in these tables are those obtained from the molecular constants determinated by the band-by-band studies. The next step, obtaining the most adequate spectroscopic constants capable of reproducing the iodine spectrum within experimental errors, was then undertaken: the first results



of this global fit and Dunham expansions of the molecular constants have been given at a recent conference (5).

3. COMPARISON WITH PREVIOUSLY PUBLISHED ASSIGNMENTS

Previously published adequately reliable assignments which can be compared with the ones obtained from our studies are available only for three of the six small

314



FIG. 5. Assignments of the iodine absorption lines in the vicinity of the Kr⁺ laser line $\lambda = 5683$ Å ($\sigma = 17$ 594 cm⁻¹).

portions given here: the region around 19 429.8 cm⁻¹ (5145 Å, Ar⁺ laser), investigated by a great number of authors [(11, 12), for example] was carefully reassigned by Tellinghuisen (13); the other two regions around 18 908 cm⁻¹ (5287 Å, Ar⁺ laser), and 18 830 cm⁻¹ (5308 Å, Kr⁺ laser) are identified in the atlas (18000– 19000 cm⁻¹) recently published by Simons and Hougen at the National Bureau of Standards (14).



3.1. The Spectral Region around 19 429.8 cm⁻¹ (5145 Å, Ar⁺ Laser)

(a) Comparison between the assignments based on the molecular constants determinated by Wei and Tellinghuisen (10), and those deduced from Fourier spectroscopy. Inspection of Table VII shows the improvements reached by the Fourier spectroscopy measurements in the region around 19 429.8 cm⁻¹. The agreement between observed and calculated wavenumbers is, in general, good especially for unperturbed single lines as, for example, the intense rotational line (44,0) R(50) [Fig. 7, Table VII (where $\sigma_{cal} - \sigma_{obs} \approx 0.001 \text{ cm}^{-1}$)]. The assignments differ

I_2 LINES IN THE B-X SYSTEM

TABLE VII

[Wavenumber σ (cm ⁻¹)				
Band	Line	Int.	Observed	Calculated			
v',v"		Obser. [1]	Fourier Spectroscory [1]	Fourier Spectrosc. [5]	Wei [10] Tel. [13]	Barrow Yee [9]	(J _{max}) [9]
46,0	r (80)	63	19429.1831	. 1837		. 17	94
43,0 43,0	P (14) R (16)	74	.2612	.2582 .2672	.27 .28	.24 .25	60
49.1	P (27)	21	.3184	.3160	.30	• 33	82
52,0	R(119)	25	.3770	.3900	.87	(30,150)	64
44,0	R (50)	72	. 4501	.4491	.46	.44	68
60,2	r (49)			.4810	.44	.56	67
51,1	r (60)	20	.5085	.5088	.48	.50	62
44,0 49,0	P (48) P(103)	77	•6064	.6040 .6156	.62 .70	.60 .71	68 82
60,2	P (48)			.6680	.63	.75	67
58,1	r (98)	12	.7382	.7359	. 94	.61	62
61,2	r (54)			.8050	.76	.84	62
43,0 43,0	P (*3) R (15)	74	.8204	.8157 .8250	.83 .84	.80 .81	60
50,1 47,0	P (46) P (88)	61	. 9343	.9267 .9362	.91 .95	.93 .91	50 99
49,1	R (28)	16	. 9997	.0020	.98	.01	82
46,0 57,1	P (78) R (95)	64	19430.0505	.0505 .0617	.06 .17	.03 .14	94 59
50,0	r (110)	34	. 1107	.1069	.31	(29.650)	50
62.2 52,1 57,1 58,1 61,2	R (58) R (69) P (94) P (97) P (53)	25	. 1491	. 138 . 146 . 151 . 176 . 184	.10 .12 .22 .32 .15	. 17 . 13 . 19 . 03 . 22	66 64 59 62 62

Comparison between Wavenumbers Calculated with Different Sets of Constants in the Vicinity of the Ar⁺ Laser Line $\lambda = 5145$ Å ($\sigma = 19$ 429.8 cm⁻¹)

slightly from those deduced from the molecular constants published by Wei and Tellinghuisen (10). For example, the line (58,1) R(98) is now correctly assigned. In this region five pairs of lines, namely, the pairs [(43,0) P(14), R(16)], [(43,0) P(13), R(15)], [(50,1) P(46), (47,0) P(88)], [(44,0) P(48), (49,0) P(103)], and [(46,0) P(78), (57,1) R(95)], coincide within 0.010 cm⁻¹ (see column 5, Table VII). Now if we consider the calculated wavenumber based on the molecular constants given by Wei and Tellinghuisen (10), one notes that the calculated distances be-



FIG. 7. Assignments and widths of the iodine absorption lines in the vicinity of the Ar⁺ laser line $\lambda = 5145$ Å (19 429.8 cm⁻¹). The distance between the two complex lines [P(48), P(103)] and [P(88), P(46)] is found to be 0.328 cm⁻¹ or ~10 GHz [see Ref. (17)].

tween the lines of each pair increase with the J values of the rotational lines involved in these pairs: for the [P(14), R(16)] and the [P(13), R(15)] pairs of lines the calculated separations are 0.010 cm⁻¹ in agreement with our estimate, but the

TABLE VIII

	Assignments of the	Doublet splittings (cm ⁻¹)			
Absorbed lines	fluorescence doublets	Fabry-Perot Data	Fourier Data		
[17]	[13][16][17]	Measured [17]	Measured [1]	Calculated [5]	
44.0 P (48)	44.0 P (48) R (46)	7.1 ± 0.3	7.085	7.085	
49,0 P(103)	49,0 P(103) R (101)	15.3 ± 0.3	15.270	15.258	
58,1 R (98)	58,1 R (98) P(100)	14.8 ± 0.3	14.767	14.782	
43, 0 P (13)	43,0 P (13) R (11)	1.87 ± 0.3	1.858	1.866	
43,0 R (15)	43,0 R (15) P (17)	2.46 ± 0.3	2.454	2.462	
47,0 p (88)	47.0 P (68) R (86)	13.1 ± 0.3	13,036	13.035	

Doublets Splittings Measurements: Comparison between Fabry-Perot Measurements (17) and Fourier Measurements

separations for the pairs [P(46), P(88)], [P(48), P(103)], and [P(78), R(95)], wouldbe 0.040, 0.080, and 0.110 cm⁻¹, respectively. [In fact, the recorded half-widthsof the spectral lines in this region (Fig. 7) have a maximum value of 0.040 cm⁻¹,of which the hyperfine structure contributes about <math>0.025-0.030 cm⁻¹ (11, 15)]. This is a good illustration of the observation made in Section 2, viz., that at the head of the bands (low J) the discrepancy is low and nearly constant, following which it increases continuously with J. Finally it can be noted that using a monomode Ar⁺ laser the rotational line (49,1) R(28) cannot be easily excited simultaneously with the composite line [(50,1) P(46), (47,0) P(88)], which is located 0.065 cm⁻¹, toward the lower wavenumbers.

(b) Comparison between the assignments based on the molecular constants determinated by Barrow and Yee (9), and the assignments deduced from Fourier spectroscopy. Other molecular constants describing the B-X system of I_2 were published by Barrow and Yee (9). The wavenumbers calculated by means of these constants, which are listed in Table 2 of Ref. (9), are given in Table VII, column 7. When the J values of the transitions are below the maximum J values identified in each band by these authors $[J_{max}$ of Table I of Ref. (9), and reproduced in column 8, Table VII of the present work], then the differences between the observed Fourier wavenumbers (column 4) and those calculated using the absorption data of Barrow and Yee (column 7) do not exceed ± 0.020 cm⁻¹. However, for J values higher than those J_{max} values, the discrepancies increase rapidly; for example, for the lines (57,1) P(94), (57,1) R(95), (58,1) P(97), (58,1) R(98), (49,0) P(103), (50,0) R(110), and (52,0) R(119) the discrepancies are -0.040, -0.090, +0.140, +0.130, -0.090, +0.460, and -0.770 cm⁻¹, respectively. Therefore, the use of the constants of Barrow and Yee has to be limited to the maximum J values of each band observed by these authors. Since the molecular constant H'_v has been neglected in their study, the extrapolation to high J values remains questionable and, in any case, has to be made with extreme caution. For instance, in the recent laser fluorescence studies [Clark and McCaffery (16)] the assignments of the fluorescence series, although correct, are given for two of them with wrong wavenumbers: the wavenumbers of the observed transitions (49,0) P(103) and (58,1)R(98) must be inverted, as can be seen by inspection of Table VII. These misassignments have, obviously, their origin in the use and the extrapolation to relatively high J values, of the data of Barrow and Yee. [Another fluorescence series (61,2) P(54) is given in Ref. (16); it should be (61,2) R(54), which is in agreement both with our assignments (Table VII), and with their own observations; see Fig. 2 of Ref. (16).]

(c) Comparison between fluorescence doublet splittings measurements by means of a Fabry-Perot interferometer and doublet measurements obtained by Fourier spectroscopy. In view of the precise nature of the data obtained by Fourier spectroscopy it is also now possible to give a comprehensive interpretation of the fluorescence doublet measurements, in this region, made by Patterson *et al.* using a Fabry-Perot interferometer (17). Table VIII contains the splittings of the six doublets observed by these authors (17) and their corresponding assignments. The agreement between the measured splittings (column 3, Table VIII) with those calculated by means of our molecular constants (5), or with those obtained from the observed wavenumbers (1) of the doublet components, is excellent.

Again, in the experiment of Patterson *et al.* (17) the separation between the lines (44,0) P(48) and (49,0) P(103) is found to be of the order of 571 MHz (0.019 cm⁻¹) in agreement with our observation; Clark and McCaffery found them to be about 0.110 cm⁻¹ apart, a value not compatible with the half-width (0.040 cm⁻¹) of the complex line P(48) P(103) recorded in our spectrum (see Fig. 7).

Finally, for completeness and to conclude the discussion of the assignments of the iodine lines in this spectral region, Table VII and Fig. 7 also contain transitions (dashed lines) connected to the "hot" band v'' = 2. These bands, very weak in our experimental conditions (1, 18), were not analyzed; only the calculated wavenumbers are given here; transitions belonging to the (v', 2) bands with v' > 62 were not searched, our assignments being limited to v' = 62.

3.2. The Spectral Regions around 18 908 cm⁻¹ (5287 Å, Ar⁺ Laser) and 18 830 cm⁻¹ (5308 Å, Kr⁺ Laser)

In these two regions the number of recorded lines by Fourier spectroscopy is about 30% higher than those recorded in the spectrum published by Simons and Hougen (14). Of course, the 30% additional lines are weak and are in general close to intense lines: the lines could be detected because of the high instrumental resolving power used, which was of the order of 10^6 .

For the (v',0) transitions, the assignments of Simons and Hougen agree completely with ours. Moreover, recently we were made aware of the Franck-Condon factors calculated by Tellinghuisen (18). Consequently, we have systematically searched in these two regions for the lines belonging to the bands (v',v'') with $28 \le v \le 46$ and v'' = 2. For these bands the Franck-Condon factors should be high enough so that transitions belonging to the bands (v',2) would be detectable in our spectrograms. Indeed, we find these bands and the identification of the corresponding lines is given in Fig. 2, Table II and Fig. 4, Table IV.

In conclusion, we wish to point out that assignments in other regions of the iodine absorption spectrum can be made equally well with the use of the molecular constants deduced (5) from our Fourier spectroscopy measurements.

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