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Basis set convergence of the interaction energy of hydrogen-bonded complexes

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The Hartree-Fock and correlation contributions to the interaction energy of the hydrogen-bonded complexes (HF)₂, (HCl)₂, H₂OHF, HCNHF, and (H₂O)₂ are computed in conventional calculations employing the aug-cc-pVXZ series of basis sets at the levels of Hartree-Fock theory, second-order perturbation theory, and coupled-cluster theory with single and double excitations augmented by a perturbative triples correction. The basis set convergence of the interaction energy is examined by comparison with results obtained with an explicitly correlated wave function model. The counterpoise-corrected and uncorrected Hartree-Fock interaction energies both converge very unsystematic because the basis set superposition error and the error from the incomplete description of the electronic Coulomb cusp both are present. Once the former has been effectively removed by the counterpoise correction, the cusp dominates and the convergence of the counterpoise-corrected correlation energy. Two-point extrapolated values obtained with this form are close to the basis set limit and represent a significant improvement on the unextrapolated results. © *1999 American Institute of Physics*. [S0021-9606(99)30844-8]

I. INTRODUCTION

The accurate determination of interaction energies of hydrogen-bonded complexes has been the subject of many theoretical studies¹⁻²¹ (the list should not be considered exhaustive) and represents an important issue of ab initio theory. When aiming at an accurate determination of the interaction energy of a hydrogen-bonded complex within the super-molecule approach, the basis set superposition error (BSSE) must be accounted for. BSSE is a spurious contribution to the interaction energy arising from the improved description of each fragment in the total basis of the complex compared to the fragment basis alone, and is thus ultimately a consequence of the incompleteness of the one-electron basis set. A conceptually simple way of accounting for BSSE is the counterpoise correction method,²² in which the energies of the fragments are calculated in the full basis of the complex, and these counterpoise-corrected (CP) energies are used for the energies of the fragments when computing the interaction energy. The accuracy of the simple counterpoise

correction as a measure of BSSE has been the subject of some debate.^{23–30} However, there appears to be a consensus that this method at least gives a correct estimate of BSSE, and as it furthermore is the procedure recommended to account for BSSE in a recent review on counterpoise theory,³⁰ we shall not dwell on this matter here and instead simply use the counterpoise-correction as our measure of BSSE.

Recognizing the incompleteness of the basis set as the origin of BSSE, the focus of the present study will be on the basis set convergence of the interaction energy toward the basis set limit, where BSSE vanishes. When studying basis set convergence, one must have a hierarchical sequence of basis sets with systematic improvements from level to level. The correlation-consistent polarized valence basis sets, denoted cc-pVXZ, and their singly and doubly diffusely augmented counterparts, denoted aug-cc-pVXZ and d-aug-cc-pVXZ, respectively, of Dunning and co-workers,^{31–34} constitute such hierarchies and were used exclusively in our first investigation of the prototypical hydrogen-bonded complex $(H_2O)_2$.¹⁷ Hydrogen-bonded complexes are weakly bound and have rather long interfragment distances. An accurate description of the outer valence region of each frag-

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ment is therefore required, and the inclusion of diffuse functions in the basis set becomes important. Accordingly, Del Bene,⁷ Feller et al.,^{8,13} and Xantheas et al.^{9,14} obtained major improvements in the interaction energy from the cc-pVXZ to the aug-cc-pVXZ basis sets, and in Ref. 17 the size of the counterpoise correction for (H₂O)₂ was found to be reduced significantly from the cc-pVXZ to the aug-cc-pVXZ basis sets, whereas little happened when the next set of diffuse functions (in the d-aug-cc-pVXZ sets) was added. Besides confirming the importance of diffuse functions, it was found in Ref. 17 that the CP interaction energies (ΔE^{CP}) were closer to the basis set limit than the uncorrected ones (ΔE) for the Hartree-Fock self-consistent field (SCF) wave function model. It was also found that ΔE^{CP} and ΔE converged to the basis set limit from each side for all of the three correlated wave function models: second-order Møller-Plesset perturbation theory (MP2),³⁵ coupled-cluster singles and doubles (CCSD),³⁶ and CCSD augmented by a perturbational correction for connected triple excitations [CCSD(T)].³⁷ Furthermore, once diffuse functions were included in the basis set, ΔE^{CP} displayed larger variations with the cardinal number X(D:2,T:3...) of the basis set than ΔE , and ΔE obtained with the small aug-cc-pVDZ basis set was remarkably close to the basis set limit. Finally, because ΔE^{CP} and ΔE for the correlated wave function models converged to the basis set limit from each side, the simple average of ΔE^{CP} and ΔE converged rapidly to the basis set limit.

In the present study, we investigate these matters in more detail for a broader sample of five hydrogen-bonded complexes, to establish whether our findings in Ref. 17 were specific for the water dimer or hold for hydrogen-bonded complexes in general. The systems under investigation $(HF)_2$, $(HCl)_2$, H_2OHF , HCNHF, and $(H_2O)_2$ contain hydrogen bonds involving N, O, F, and Cl, and thus cover most of the hydrogen bonds encountered in nature. In Ref. 17, the total interaction energy for the correlated wave function models was examined. In this study, we instead focus on each of the SCF and correlation contributions to the total interaction energy separately, as these contributions converge differently to the basis set limit.

Explicitly correlated linear R12-methods³⁸⁻⁴³ give correlation energies close to the basis set limit, because they include terms in the wave function that are linear in the interelectronic distance r_{12} explicitly, as required by the electronic Coulomb cusp condition, and these methods have been successfully applied for hydrogen-bonded systems.^{12,19-21} Guided by basis set limits established with the MP2-R12/A method, the basis set convergence of the MP2 correlation contribution to the interaction energy is investigated in detail in the present study. The convergence of $\Delta E_{\rm corr}$, $\Delta E_{\rm corr}^{\rm CP}$, and their simple average is examined for the aug-cc-pVXZ basis sets, and the performance of the recently advanced two-point X^{-3} extrapolation for acceleration of basis set convergence⁴⁴⁻⁴⁷ is investigated. Finally, based on the experience obtained at the MP2 level, the basis set convergence of the CCSD(T) correlation contribution is investigated.

II. COMPUTATIONAL DETAILS

For consistency with our previous work on the water dimer, the geometries of all the complexes have been optimized at the CCSD(T)/aug-cc-pVTZ level with the monomer fragments frozen at the experimental equilibrium geometry— $R_{\text{H-F}}=0.916\,80,^{48}$ $R_{\text{H-CI}}=1.274\,55,^{48}$ $R_{\text{H-O}}=0.9572\,\text{\AA}$ and $\angle_{\text{H-O-H}}=104.52^{\circ},^{49}$ and $R_{\text{H-C}}=1.062\,55$ and $R_{\text{C-N}}=1.152\,87\,\text{\AA}.^{50}$ All the electrons were correlated in the geometry optimizations, except for (HCl)₂, where the ten lowest lying molecular orbitals—essentially the 1*s*, 2*s*, and 2*p* orbitals for the two chlorine—were kept uncorrelated, as the aug-cc-pVTZ basis set does not provide a sufficiently flexible description of the inner valence and core region around chlorine for correlating these orbitals. These calculations have been performed with the ACESII program.^{51,52}

Conventional SCF, MP2, and CCSD(T) single-point energy calculations at the optimized geometries have been carried out for the aug-cc-pVXZ basis sets (X=D-5) with only the valence electrons correlated. These calculations have been performed with a local version of the Dalton program⁵³ that contains the coupled-cluster code of Koch and co-workers.⁵⁴⁻⁵⁶ The explicitly correlated MP2-R12/A single-point energy calculations have been carried out with the SORE program⁵⁷ correlating only the valence electrons. Large uncontracted basis sets of the form 21s13p10d8f (Cl), 15s9p7d5f (C, N, O, F), and 9s7p5d (H) have been employed to satisfy completeness conditions related to the approximate evaluation of the many-electron integrals entering the MP2-R12 theory. These basis sets are constructed from the primitive spdf set of the cc-pV5Z basis set by addition of both diffuse and tight functions. The sets for H, N, O, and F are described in full detail in Ref. 58. For C, the primitive 14s8p4d3f set in the cc-pV5Z basis set is augmented by a 1s1p1d1f set of diffuse functions with exponents 0.0408 (s), 0.0308 (p), 0.075 (d), and 0.146 (f), and with a 2d1f set of tight functions with exponents 20.25 (d), 7.966 (d), and 4.8 (f), and for Cl, the primitive 20s12p4d3f set in the cc-pV5Z basis set is augmented by a 1s1p1d1f set of diffuse functions with exponents 0.0537 (s), 0.0369 (p), 0.101(d), and 0.156 (f), and with a 5d4f set of tight functions with exponents 350.0 (d), 141.4 (d), 57.17 (d), 23.12 (d), 9.35 (d), 23.9 (f), 11.6 (f), 5.66 (f), and 2.76 (f).

From the calculated energies we obtain the uncorrected interaction energy as

$$\Delta E = E^{\text{comp}} - E^{\text{don}} - E^{\text{acc}},\tag{1}$$

where E^{comp} is the energy of the complex, and E^{don} and E^{acc} are the energies of the isolated H donor and acceptor monomers, respectively. The CP interaction energy is obtained as

$$\Delta E^{\rm CP} = E^{\rm comp} - E^{\rm CP,don} - E^{\rm CP,acc},\tag{2}$$

where $E^{\text{CP,don}}$ and $E^{\text{CP,acc}}$ are the energies of the two monomers in the full basis of the complex. The correlation contributions, ΔE_{corr} and $\Delta E_{\text{corr}}^{\text{CP}}$, to the interaction energy are obtained from analogous expressions involving the corresponding correlation energies.

TABLE I. Cartesian coordinates for the CCSD(T)/aug-cc-pVTZ optimized equilibrium geometries of the complexes in atomic units.

Atom	X	Y	Ζ
$(HF)_2$			
Н	3.214 605 13	-1.49640178	0.000 000 00
F	2.511 286 53	0.086 918 66	$0.000\ 000\ 00$
Н	-0.91296787	0.186 832 05	0.000 000 00
F	$-2.633\ 383\ 52$	$-0.017\ 448\ 75$	0.000 000 00
$(HCl)_2$			
Н	3.498 634 66	$-2.343\ 865\ 35$	0.000 000 00
Cl	3.548 559 49	0.064 168 10	0.000 000 00
Н	-1.225 351 64	0.303 046 86	0.000 000 00
Cl	-3.61407698	$-0.005\ 350\ 39$	0.000 000 00
H_2OHF			
Н	0.712 704 86	-3.290 732 19	1.430 432 08
Н	0.712 704 86	-3.290 732 19	-1.43043208
0	$-0.101\ 818\ 48$	$-2.540\ 829\ 48$	0.000 000 00
Н	$-0.068\ 203\ 40$	0.719 544 76	0.000 000 00
F	0.013 724 89	2.450 107 80	0.000 000 00
HCNHF			
Н	0.000 000 00	$0.000\ 000\ 00$	$-5.722\ 109\ 90$
С	0.000 000 00	$0.000\ 000\ 00$	$-3.714\ 180\ 98$
Ν	0.000 000 00	$0.000\ 000\ 00$	-1.535 571 97
Н	0.000 000 00	$0.000\ 000\ 00$	1.945 645 75
F	0.000 000 00	$0.000\ 000\ 00$	3.678 147 03
$(H_2O)_2^{a}$			
Н	-3.30256970	$-0.785\ 537\ 00$	1.430 432 10
Н	-3.30256970	$-0.785\ 537\ 00$	-1.43043210
0	$-2.666\ 109\ 50$	0.120 399 50	0.000 000 00
Н	1.006 003 70	0.105 080 40	0.000 000 00
Н	3.473 430 20	1.552 958 40	0.000 000 00
0	2.800 048 30	-0.125 879 10	0.000 000 00

III. RESULTS AND DISCUSSION

The Cartesian coordinates of the CCSD(T)/aug-cc-pVTZ optimized equilibrium geometries of the complexes are listed in Table I, and in Tables II through V, the single-point energies and the interaction energies deduced from these are given for the SCF, MP2, and CCSD(T) wave function models.

A. Convergence in Hartree–Fock calculations

From the SCF results in Table II, we note that ΔE and ΔE^{CP} approach each other when X increases, as they should. Although ΔE and ΔE^{CP} approach each other, the basis set convergence of both is very unsystematic, and it is difficult to find features common to all five complexes. This is further illustrated in Figs. 1–5, where ΔE and ΔE^{CP} at the SCF level are plotted as a function of X. The convergence is very uneven and on several occasions it is nonmonotonic for ΔE^{CP} . Furthermore, ΔE^{CP} generally converges from above, except for (HCl)₂, where ΔE^{CP} is virtually constant with X, but the convergence differs significantly from complex to complex, with the largest reductions in the basis set error occurring at different X. By contrast, ΔE converges from below in all five cases, but again the detailed convergence differs from complex.

At the aug-cc-pV5Z level, ΔE and ΔE^{CP} differ only by about 0.01 mE_h, and their average [$\Delta E^{ave}(5)$] is therefore expected to be within a few hundredths of a mE_h of the SCF basis set limit. For the comparisons to be made in the fol-

TABLE II. The SCF results obtained for the five complexes as a function of the cardinal number X of the aug-cc-pVXZ basis sets. The first column of results holds the energy of the complex, the second column the energy of the isolated H donor monomer, the third column the energy of the isolated H acceptor monomer, and in the next two columns follow the energy of these two monomers in the full basis set of the complex. The final three columns hold the uncorrected interaction energy, the CP interaction energy, and the simple average of these two. All the interaction energies are given in mE_h , whereas the others are given in E_h .

X	$E^{\rm comp}$	E^{don}	E^{acc}	$E^{\rm CP,don}$	$E^{\rm CP,acc}$	ΔE	ΔE^{CP}	$\Delta E^{ m ave}$	
(HF) ₂									
2	-200.073 147	-100.033474	-100.033474	-100.033 580	-100.033768	-6.1991	-5.7991	-5.9991	
3	$-200.128\ 141$	-100.061078	-100.061078	-100.061 125	-100.061 220	-5.9854	-5.7959	-5.8906	
4	-200.143143	-100.068568	-100.068568	-100.068594	$-100.068\ 630$	-6.0059	-5.9181	-5.9620	
5	$-200.147\ 103$	-100.070583	-100.070583	-100.070589	-100.070589	-5.9382	-5.9255	-5.9318	
				(HCl) ₂					
2	-920.185955	$-460.092\ 615$	$-460.092\ 615$	$-460.092\ 691$	-460.092845	-0.7245	-0.4193	-0.5719	
3	-920.215 674	$-460.107\ 600$	$-460.107\ 600$	$-460.107\ 624$	$-460.107\ 659$	-0.4743	-0.3909	-0.4326	
4	$-920.223\ 209$	-460.111 395	-460.111 395	-460.111 403	$-460.111\ 414$	-0.4192	-0.3923	-0.4057	
5	$-920.225\ 400$	-460.112504	-460.112504	-460.112508	-460.112 509	-0.3918	-0.3834	-0.3876	
				H ₂ OHF					
2	-176.086740	-100.033474	-76.041428	$-100.033\ 612$	-76.041 822	-11.8379	-11.3053	-11.5716	
3	-176.133 307	$-100.061\ 078$	$-76.060\ 613$	$-100.061\ 145$	$-76.060\ 708$	-11.6163	-11.4542	-11.5353	
4	$-176.146\ 150$	-100.068568	$-76.066\ 001$	$-100.068\ 612$	-76.066037	-11.5809	-11.5011	-11.5410	
5	-176.149415	-100.070583	-76.067 320	-100.070591	-76.067 324	-11.5121	-11.4994	-11.5057	
				HCNHF					
2	$-192.930\ 371$	-100.033474	-92.887498	$-100.033\ 632$	$-92.887\ 888$	-9.3986	-8.8509	-9.1248	
3	-192.979 645	$-100.061\ 078$	-92.909 182	$-100.061\ 160$	-92.909 323	-9.3852	-9.1619	-9.2735	
4	-192.992434	-100.068568	-92.914 635	$-100.068\ 629$	-92.914 661	-9.2311	-9.1448	-9.1879	
5	-192.995 518	-100.070583	-92.915778	-100.070593	-92.915781	-9.1569	-9.1439	-9.1504	
				$(H_2O)_2$					
2^{a}	$-152.088\ 835$	-76.041428	-76.041428	-76.041 511	-76.041 747	-5.9797	-5.5776	-5.7787	
3 ^a	-152.126908	$-76.060\ 613$	$-76.060\ 613$	$-76.060\ 660$	$-76.060\ 689$	-5.6817	-5.5585	-5.6201	
4^{a}	-152.137 667	-76.066001	-76.066001	-76.066016	-76.066033	-5.6657	-5.6175	-5.6416	
5 ^a	-152.140 264	-76.067 320	-76.067 320	-76.067 323	-76.067 324	-5.6234	-5.6168	-5.6201	



FIG. 1. ΔE^{CP} (upper curve) and ΔE (lower curve) for (HF)₂ at the SCF level as a function of the cardinal number *X*. The unconnected diamonds (\diamond) are the simple average of the ΔE^{CP} and ΔE results.

lowing, this accuracy is sufficient, and we shall therefore adopt $\Delta E^{\text{ave}}(5)$ as our estimate of the SCF basis set limit. We then observe that, at the aug-cc-pVQZ level, ΔE^{CP} is always closer to the basis set limit than the corresponding ΔE is, and except for two cases this also holds for the smaller basis sets. The simple average ΔE^{ave} represents an improvement on the corresponding ΔE for the double-, triple-, and quadruple-zeta basis sets, and in many cases the improvement is substantial. Compared to ΔE^{CP} , however, ΔE^{ave} does not always represent an improvement and it is not possible to recommend any one of these two in favor of the other. The largest basis set errors are observed for ΔE at the aug-cc-pVDZ level, where the error is about 0.3 mE_h ≈ 0.2 kcal/mol. At the aug-cc-pVTZ level, both ΔE^{CP} and $\Delta E^{\rm ave}$ are within 0.1 kcal/mol of the basis set limit, and at the aug-cc-pVQZ level, $\Delta E^{\rm CP}$ and $\Delta E^{\rm ave}$ are both within 0.05 kcal/mol of the basis set limit.

Because of the very unsystematic convergence it is not possible to find an analytical form for the convergence of ΔE or ΔE^{CP} at the SCF level, and reliable extrapolation schemes cannot be developed.

B. Convergence of the MP2 correlation contribution

The MP2 correlation contributions in Table III show that ΔE_{corr} and $\Delta E_{\text{corr}}^{\text{CP}}$ also approach each other when X in-



FIG. 3. ΔE^{CP} (upper curve) and ΔE (lower curve) for H₂OHF at the SCF level as a function of the cardinal number *X*. The unconnected diamonds (\diamond) are the simple average of the ΔE^{CP} and ΔE results.

creases, as they should. However, for a given X the differences between the two results are much larger for the correlation part than for the SCF part (e.g., about 0.2 to 0.3 mE_h compared to about 0.01 mE_h at the aug-cc-pV5Z level). The R12 results for ΔE_{corr} and ΔE_{corr}^{CP} are much closer to each other, differing by more than 0.04 mE_h only for (HCl)₂. The results are depicted in Fig. 6–10, where ΔE_{corr} and ΔE_{corr}^{CP} for the aug-cc-pVXZ basis sets are plotted as a function of X along with the two R12 results (CP and uncorrected interaction energies). The basis set limit is expected to lie close to these two R12 results, and their range indicates the uncertainty of its determination.

The convergence of $\Delta E_{\rm corr}^{\rm CP}$ is more systematic than that of $\Delta E^{\rm CP}$ at the SCF level: the five curves for $\Delta E_{\rm corr}^{\rm CP}$ look exactly alike, and the convergence of $\Delta E_{\rm corr}^{\rm CP}$ is both monotonic, smooth, and invariably from above. The convergence of $\Delta E_{\rm corr}$, however, is still unsystematic, although some common features are noted for all five complexes: The convergence is always nonmonotonic, as $\Delta E_{\rm corr}$ invariably decreases from aug-cc-pVDZ to aug-cc-pVTZ and increases from aug-cc-pVTZ to aug-cc-pVDZ to aug-cc-pVTZ converges from below, although not particularly smoothly. Especially, the decrease from aug-cc-pVDZ to aug-cc-pVTZ causes the aug-cc-pVTZ $\Delta E_{\rm corr}$ results to be *further away* from the basis set limit than the aug-cc-pVDZ $\Delta E_{\rm corr}$ results.



FIG. 2. ΔE^{CP} (upper curve) and ΔE (lower curve) for (HCl)₂ at the SCF level as a function of the cardinal number *X*. The unconnected diamonds (\diamond) are the simple average of the ΔE^{CP} and ΔE results.



FIG. 4. ΔE^{CP} (upper curve) and ΔE (lower curve) for HCNHF at the SCF level as a function of the cardinal number *X*. The unconnected diamonds (\diamond) are the simple average of the ΔE^{CP} and ΔE results.



FIG. 5. ΔE^{CP} (upper curve) and ΔE (lower curve) for (H₂O)₂ at the SCF level as a function of the cardinal number *X*. The unconnected diamonds (\diamond) are the simple average of the ΔE^{CP} and ΔE results.

This behavior may be attributed to the remarkably good performance of the aug-cc-pVDZ basis set, which must arise from a fortuitous cancellation of errors between BSSE and the other (ordinary) basis set errors (e.g., those arising from the incomplete description of the electronic Coulomb cusp). The $\Delta E_{\rm corr}^{\rm CP}$ results for the aug-cc-pVDZ and aug-cc-pVTZ basis sets have larger basis set errors than the corresponding $\Delta E_{\rm corr}$ results. Especially the $\Delta E_{\rm corr}^{\rm CP}$ aug-cc-pVDZ results are very poor, with basis set errors from 0.8 to $1.3 \,\mathrm{mE}_h \approx 0.5$ to 0.8 kcal/mol, while the errors of the $\Delta E_{\mathrm{corr}}^{\mathrm{CP}}$ aug-cc-pVTZ results are between 0.2 and 0.4 kcal/mol. As ΔE_{corr} and $\Delta E_{\mathrm{corr}}^{\mathrm{CP}}$, beyond the double-zeta level, converge toward the basis set limit from each side, their average is close to the basis set limit. At the aug-cc-pVTZ level, the average is within 0.15 kcal/mol of the basis set limit, whereas at both the aug-cc-pVQZ and aug-cc-pV5Z levels it is within 0.05 kcal/mol of this limit. However, it should be noted that in three cases, $\Delta E_{\mathrm{corr}}^{\mathrm{ave}}$ at the aug-cc-pV5Z level is slightly farther from the basis set limit than the $\Delta E_{\mathrm{corr}}^{\mathrm{ave}}$ aug-cc-pVQZ result is, and in all these cases the aug-cc-pV5Z average result is slightly too low.

C. Extrapolations of the MP2 correlation contribution

The basis set convergence of the correlation energy for the correlation-consistent basis sets has recently been found to follow the form^{44,45}

$$E_{\operatorname{corr},X} = E_{\operatorname{corr},\lim} + AX^{-3},\tag{3}$$

where $E_{\text{corr},X}$ is the correlation energy obtained with the correlation-consistent basis set with cardinal number *X*, and $E_{\text{corr,lim}}$ is the basis set limit value of the correlation energy. For large *X*, Eq. (3) displays the correct asymptotic behavior of the correlation energy^{44–46} compared with the theoretical

TABLE III. The MP2 correlation contribution to the different results obtained for the five complexes as a function of the cardinal number X of the aug-cc-pVXZ basis sets. The ordering of the columns of results is the same as the one in Table II, and the rows with R12 for X give the MP2-R12 results. All the interaction anergy contributions are given in mE_h , whereas the others are given in E_h .

X	$E_{\rm corr}^{\rm comp}$	$E_{\rm corr}^{ m don}$	$E_{ m corr}^{ m acc}$	$E_{\rm corr}^{ m CP,don}$	$E_{\rm corr}^{ m CP,acc}$	$\Delta E_{ m corr}$	$\Delta E_{ m corr}^{ m CP}$	$\Delta E_{ m corr}^{ m ave}$
				$(HF)_2$				
2	-0.445563	-0.222260	-0.222260	-0.222 569	-0.222676	-1.0440	-0.3186	-0.6813
3	-0.560927	-0.279785	-0.279785	-0.280028	-0.280138	-1.3568	-0.7604	-1.0586
4	$-0.603\ 657$	-0.301 189	-0.301 189	-0.301 318	-0.301 391	-1.2802	-0.9482	-1.1142
5	-0.621218	-0.309974	-0.309974	-0.310086	$-0.310\ 115$	-1.2703	-1.0170	-1.1436
R12	-0.642992	-0.320947	-0.320947	-0.320962	-0.320918	-1.0990	-1.1124	-1.1057
				(HCl) ₂				
2	-0.321260	-0.159096	-0.159096	-0.159391	-0.159659	-3.0666	-2.2088	-2.6377
3	-0.418444	-0.207530	-0.207530	-0.207721	$-0.207\ 901$	-3.3829	-2.8217	-3.1023
4	-0.452737	-0.224704	-0.224704	-0.224812	-0.224918	-3.3303	-3.0077	-3.1690
5	$-0.467\ 861$	$-0.232\ 273$	-0.232273	-0.232357	-0.232428	-3.3154	-3.0758	-3.1956
R12	-0.493075	-0.244997	-0.244997	$-0.245\ 001$	-0.244884	-3.0803	-3.1894	-3.1349
				H ₂ OHF				
2	-0.443550	-0.222260	-0.219337	-0.222729	$-0.220\ 197$	-1.9531	-0.6233	-1.2882
3	$-0.550\ 285$	-0.279785	$-0.268\ 345$	$-0.280\ 121$	$-0.268\ 802$	-2.1546	-1.3617	-1.7581
4	$-0.589\ 290$	$-0.301\ 189$	$-0.285\ 912$	-0.301407	-0.286191	-2.1893	-1.6919	-1.9406
5	$-0.605\ 007$	-0.309974	-0.292905	-0.310132	$-0.293\ 062$	-2.1280	-1.8129	-1.9704
R12	-0.624219	-0.320947	-0.301 323	-0.320967	$-0.301\ 277$	-1.9491	-1.9754	-1.9622
				HCNHF				
2	-0.519571	-0.222260	-0.294213	-0.222758	-0.294904	-3.0989	-1.9095	-2.5042
3	$-0.633\ 111$	-0.279785	$-0.350\ 167$	-0.280144	-0.350557	-3.1591	-2.4098	-2.7845
4	-0.674288	$-0.301\ 189$	-0.369877	-0.301444	-0.370188	-3.2224	-2.6553	-2.9389
5	-0.690684	-0.309974	-0.377664	-0.310142	-0.377799	-3.0461	-2.7419	-2.8940
R12	-0.711878	-0.320947	-0.388044	-0.320969	$-0.388\ 053$	-2.8878	-2.8568	-2.8723
				$(H_2O)_2$				
2 ^a	-0.440914	-0.219337	-0.219337	-0.219668	-0.219972	-2.2409	-1.2743	-1.7576
3 ^a	$-0.539\ 146$	$-0.268\ 345$	$-0.268\ 345$	$-0.268\ 617$	-0.268710	-2.4557	-1.8191	-2.1374
4^{a}	$-0.574\ 185$	-0.285912	$-0.285\ 912$	-0.286024	-0.286130	-2.3613	-2.0311	-2.1962
5 ^a	$-0.588\ 123$	-0.292905	-0.292905	-0.292982	$-0.293\ 032$	-2.3134	-2.1088	-2.2111
R12 ^b	-0.604844	-0.301 323	-0.301 323	-0.301 335	-0.301 307	-2.1974	-2.2019	-2.1997

^aFrom Ref. 17. ^bFrom Ref. 21.



FIG. 6. $\Delta E_{\rm corr}^{\rm CP}$ (upper curve) and $\Delta E_{\rm corr}$ (lower curve) for (HF)₂ at the MP2 level as a function of the cardinal number *X*. The unconnected diamonds (\diamond) are the simple average of the $\Delta E_{\rm corr}^{\rm CP}$ and $\Delta E_{\rm corr}$ results, the unconnected crosses (\times) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, the asterisks (*) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, and the two horizontal dashed lines are the two R12 results.

analysis of the partial-wave expansion of the helium atom.⁵⁹ The form in Eq. (3) is therefore theoretically motivated, and extrapolations based upon Eq. (3) yield correlation energies in good agreement with those obtained from large R12 calculations.^{44,45} The fewer results obtained with basis sets of low X that are included in the extrapolations, the better the extrapolated basis set limits agree with the R12 results. Indeed, the most accurate extrapolated results beyond a given X are obtained by using only the results for X and X-1, for which the extrapolation may be cast in the following simple closed form:⁴⁵

$$E_{\text{corr,lim}} = \frac{X^3}{X^3 - (X-1)^3} E_{\text{corr,}X} - \frac{(X-1)^3}{X^3 - (X-1)^3} E_{\text{corr,}X-1}.$$
(4)

As the correlation contribution to the interaction energy is a simple linear combination of correlation energies, a form similar to Eq. (3) is expected for the correlation contribution



FIG. 7. $\Delta E_{\rm corr}^{\rm CP}$ (upper curve) and $\Delta E_{\rm corr}$ (lower curve) for (HCl)₂ at the MP2 level as a function of the cardinal number X. The unconnected diamonds (\diamond) are the simple average of the $\Delta E_{\rm corr}^{\rm CP}$ and $\Delta E_{\rm corr}$ results, the unconnected crosses (\times) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, the asterisks (*) are the two-point X^{-3} extrapolated results based results based on the $\Delta E_{\rm corr}$ results, and the two horizontal dashed lines are the two R12 results.



FIG. 8. $\Delta E_{\rm corr}^{\rm CP}$ (upper curve) and $\Delta E_{\rm corr}$ (lower curve) for H₂OHF at the MP2 level as a function of the cardinal number X. The unconnected diamonds (\Diamond) are the simple average of the $\Delta E_{\rm corr}^{\rm CP}$ and $\Delta E_{\rm corr}$ results, the unconnected crosses (×) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, the asterisks (*) are the two-point X^{-3} extrapolated results based results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, and the two horizontal dashed lines are the two R12 results.

to the interaction energy. In the following, we therefore consider two-point extrapolations for two consecutive X of the form (4).

The basis set convergence of $\Delta E_{\rm corr}^{\rm CP}$ is, as opposed to that of $\Delta E_{\rm corr}$, systematic. Furthermore, the convergence of $\Delta E_{\rm corr}^{\rm CP}$ is both monotonic and slow; closely resembling the basis set convergence of the correlation energy. The CP interaction energies therefore appear better suited for extrapolations than the uncorrected interaction energies. This is confirmed by the results in Table IV and in Fig. 6-10. As the aug-cc-pVTZ ΔE_{corr} results are further away from the basis set limit than the corresponding aug-cc-pVDZ results are, the 2–3 extrapolated ΔE_{corr} results are poorer than the original aug-cc-pVTZ ΔE_{corr} results. Likewise, the 3–4 extrapolated $\Delta E_{\rm corr}$ results do not represent much of an improvement on the original aug-cc-pVQZ ΔE_{corr} results (for H₂OHF and HCNHF they are in fact worse). Finally, except for HCNHF, the 4–5 extrapolated ΔE_{corr} results are not significantly better than the original aug-cc-pV5Z ΔE_{corr} results. This behav-



FIG. 9. $\Delta E_{\rm corr}^{\rm CP}$ (upper curve) and $\Delta E_{\rm corr}$ (lower curve) for HCNHF at the MP2 level as a function of the cardinal number X. The unconnected diamonds (\diamond) are the simple average of the $\Delta E_{\rm corr}^{\rm CP}$ and $\Delta E_{\rm corr}$ results, the unconnected crosses (\times) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, the asterisks (*) are the two-point X^{-3} extrapolated results based results based on the $\Delta E_{\rm corr}$ results, and the two horizontal dashed lines are the two R12 results.



FIG. 10. $\Delta E_{\rm corr}^{\rm CP}$ (upper curve) and $\Delta E_{\rm corr}$ (lower curve) for $({\rm H_2O})_2$ at the MP2 level as a function of the cardinal number X. The unconnected diamonds (\diamond) are the simple average of the $\Delta E_{\rm corr}^{\rm CP}$ and $\Delta E_{\rm corr}$ results, the unconnected crosses (\times) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, the asterisks (*) are the two-point X^{-3} extrapolated results based results based on the $\Delta E_{\rm corr}$ results, and the two horizontal dashed lines are the two R12 results.

ior, however, does not imply that the two-point X^{-3} extrapolation is in any way deficient. Instead, it mirrors the unsystematic convergence of $\Delta E_{\rm corr}$, which is a consequence of the combination of the basis set incompleteness errors arising from the incomplete description of the electronic Coulomb cusp and the BSSE. The X^{-3} extrapolation, by construction, remedies only the former, and as long as a substantial BSSE persists, the extrapolated values will be contaminated by this. Indeed, once the BSSE has been effectively removed by the counterpoise correction, the convergence of the correlation contribution to the interaction energy resembles that of the correlation energy itself, and the X^{-3} extrapolation gives solid improvements on the original unextrapolated $\Delta E_{\rm corr}^{\rm CP}$ results. The 2–3 extrapolated $\Delta E_{\rm corr}^{\rm CP}$ results are thus within 0.2 kcal/mol of the basis set limit, and the 3–4 extrapolated $\Delta E_{\rm corr}^{\rm CP}$ results are within 0.05 kcal/mol of the basis set limit. The 4–5 extrapolated $\Delta E_{\rm corr}^{\rm CP}$ results are also within 0.05 kcal/mol of the basis set limit, and, if any-thing, slightly above the basis set limit.

The performance of the two-point X^{-3} extrapolation for interaction energies is analogous to what has recently been observed for electric dipole moments.⁴⁷ Because of the linear dependence of the electric dipole operator on the electronic coordinates, an accurate description of the outer valence region is needed for the dipole moment, and the absence of diffuse functions in the basis set may lead to large basis set errors not related to the incomplete description of the electronic Coulomb cusp.^{47,60} Such errors are not recovered by the two-point X^{-3} extrapolation. Accordingly, when applied to the cc-pVXZ basis sets, which do not include diffuse functions, the extrapolation gave results of varying quality for the dipole moment.⁴⁷ However, when applied to the aug-ccpVXZ basis sets, dipole moments in very good agreement with basis set limits determined by R12 methods were obtained. Thus, once the significant basis set errors other than those originating from the description of the electronic Coulomb cusp have been addressed, the cusp dominates and the X^{-3} form for the basis set convergence is observed. In essence the same is seen here: once the errors arising from BSSE have been effectively removed by the counterpoise

TABLE IV. The MP2 correlation contribution to the different results obtained for the five complexes at the 2-3, 3-4, and 4-5 X^{-3} extrapolated levels and at the MP2-R12 level. The ordering of the columns of results is the same as the one in Table II. All the interaction energy contributions are given in mE_h, whereas the others are given in E_h.

X	$E_{\rm corr}^{ m comp}$	$E_{ m corr}^{ m don}$	$E_{ m corr}^{ m acc}$	$E_{\rm corr}^{ m CP,don}$	$E_{\rm corr}^{ m CP,acc}$	$\Delta E_{ m corr}$	$\Delta E_{ m corr}^{ m CP}$	$\Delta E_{ m corr}^{ m ave}$
				(HF) ₂				
2-3	$-0.609\ 501$	-0.304006	-0.304006	-0.3042222	-0.304 332	-1.4885	-0.9464	-1.2175
3-4	-0.634839	-0.316807	-0.316807	-0.316854	-0.316 900	-1.2244	-1.0851	-1.1548
4-5	-0.639643	-0.319 191	-0.319 191	-0.319285	-0.319 268	-1.2599	-1.0892	-1.1745
R12	-0.642992	-0.320947	-0.320947	-0.320962	-0.320918	-1.0990	-1.1124	-1.1057
				$(HCl)_2$				
2-3	-0.459363	-0.227924	-0.227924	$-0.228\ 070$	$-0.228\ 213$	-3.5160	-3.0798	-3.2979
3-4	-0.477763	$-0.237\ 235$	-0.237235	-0.237284	$-0.237\ 335$	-3.2920	-3.1434	-3.2177
4-5	-0.483729	-0.240214	-0.240214	-0.240273	$-0.240\ 309$	-3.2998	-3.1472	-3.2235
R12	-0.493075	-0.244997	-0.244997	$-0.245\ 001$	$-0.244\ 884$	-3.0803	-3.1894	-3.1349
				H_2OHF				
2-3	$-0.595\ 226$	$-0.304\ 006$	-0.288980	-0.304286	-0.289268	-2.2394	-1.6726	-1.9560
3-4	-0.617753	-0.316807	-0.298731	-0.316940	$-0.298\ 880$	-2.2146	-1.9329	-2.0738
4-5	-0.621497	-0.319 191	-0.300242	-0.319287	$-0.300\ 270$	-2.0637	-1.9398	-2.0017
R12	-0.624219	-0.320947	-0.301 323	-0.320967	$-0.301\ 277$	-1.9491	-1.9754	-1.9622
				HCNHF				
2-3	-0.680917	$-0.304\ 006$	-0.373726	$-0.304\ 306$	-0.373990	-3.1845	-2.6204	-2.9025
3-4	-0.704336	-0.316807	-0.384260	-0.316987	-0.384514	-3.2685	-2.8345	-3.0515
4-5	$-0.707\ 886$	-0.319 191	-0.385833	-0.319268	-0.385785	-2.8612	-2.8327	-2.8470
R12	-0.711878	-0.320947	-0.388044	-0.320969	$-0.388\ 053$	-2.8878	-2.8568	-2.8723
				$(H_2O)_2$				
2-3	-0.580507	-0.288980	-0.288980	-0.289231	$-0.289\ 227$	-2.5461	-2.0485	-2.2973
3-4	-0.599754	-0.298731	-0.298731	-0.298727	$-0.298\ 842$	-2.2924	-2.1858	-2.2391
4-5	-0.602747	-0.300242	-0.300242	-0.300283	-0.300273	-2.2631	-2.1903	-2.2267
R12 ^a	-0.604844	-0.301 323	-0.301 323	-0.301 335	-0.301 307	-2.1974	-2.2019	-2.1997

TABLE V. The CCSD(T) correlation contribution to the different results obtained for the five complexes as a function of the cardinal number X of the aug-cc-pVXZ basis sets. The ordering of the columns of results is the same as the one in Table II. The rows with 4–5, 3–4, and 2–3 for X give the results obtained at the 4–5, 3–4, and 2–3 X^{-3} extrapolated levels, respectively. All the interaction energy contributions are given in \mathbb{E}_h , whereas the others are given in E_h .

X	$E_{\rm corr}^{\rm comp}$	$E_{\rm corr}^{\rm don}$	$E_{\rm corr}^{\rm acc}$	$E_{\rm corr}^{\rm CP,don}$	$E_{\rm corr}^{\rm CP,acc}$	$\Delta E_{\rm corr}$	$\Delta E_{\mathrm{corr}}^{\mathrm{CP}}$	$\Delta E_{ m corr}^{ m ave}$
				$(HF)_2$				
2	-0.461501	-0.230 109	-0.230109	-0.230 491	-0.230622	-1.2841	-0.3883	-0.8362
3	-0.578542	-0.288480	-0.288480	-0.288741	$-0.288\ 861$	-1.5823	-0.9395	-1.2609
4	-0.619085	$-0.308\ 815$	$-0.308\ 815$	-0.308933	-0.308991	-1.4551	-1.1613	-1.3082
5	-0.633515	-0.316 049	-0.316049	-0.316 138	-0.316152	-1.4165	-1.2251	-1.3208
4-5	$-0.648\ 654$	-0.323 639	-0.323 639	-0.323697	-0.323665	-1.3760	-1.2920	-1.3340
3-4	$-0.648\ 670$	-0.323654	-0.323654	-0.323~667	-0.323680	-1.3624	-1.3232	-1.3428
2-3	$-0.627\ 822$	-0.313057	-0.313057	-0.313267	-0.313383	-1.7078	-1.1716	-1.4397
				(HCl) ₂				
2	-0.361 749	-0.179498	-0.179498	-0.179834	-0.180156	-2.7526	-1.7601	-2.2563
3	$-0.474\ 210$	$-0.235\ 630$	$-0.235\ 630$	$-0.235\ 816$	-0.235994	-2.9490	-2.3994	-2.6742
4	$-0.508\ 380$	-0.252766	-0.252766	-0.252855	-0.252926	-2.8472	-2.5993	-2.7232
5	-0.520393	-0.258787	-0.258787	$-0.258\ 847$	-0.258890	-2.8181	-2.6555	-2.7368
4-5	-0.532996	$-0.265\ 104$	$-0.265\ 104$	$-0.265\ 135$	$-0.265\ 147$	-2.7875	-2.7146	-2.7510
3-4	$-0.533\ 315$	$-0.265\ 271$	$-0.265\ 271$	$-0.265\ 288$	$-0.265\ 281$	-2.7729	-2.7451	-2.7590
2-3	-0.521562	$-0.259\ 265$	-0.259265	-0.259388	$-0.259\ 505$	-3.0317	-2.6685	-2.8501
				H_2OHF				
2	-0.464468	$-0.230\ 109$	$-0.232\ 313$	-0.230676	-0.233263	-2.0471	-0.5301	-1.2886
3	-0.572416	$-0.288\ 480$	-0.281676	$-0.288\ 840$	$-0.282\ 122$	-2.2600	-1.4542	-1.8571
4	$-0.608\ 642$	$-0.308\ 815$	-0.297581	-0.309014	-0.297799	-2.2459	-1.8282	-2.0371
5	-0.621187	-0.316 049	-0.302973	-0.316174	$-0.303\ 077$	-2.1653	-1.9362	-2.0508
4-5	-0.634349	-0.323639	$-0.308\ 629$	-0.323686	$-0.308\ 614$	-2.0808	-2.0494	-2.0651
3-4	$-0.635\ 077$	-0.323654	$-0.309\ 187$	-0.323736	$-0.309\ 240$	-2.2356	-2.1011	-2.1683
2-3	-0.617868	-0.313057	-0.302461	$-0.313\ 331$	-0.302694	-2.3497	-1.8433	-2.0965
				HCNHF				
2	-0.548~702	$-0.230\ 109$	-0.315840	-0.230694	-0.316572	-2.7540	-1.4365	-2.0953
3	$-0.663\ 181$	$-0.288\ 480$	-0.371860	$-0.288\ 865$	-0.372219	-2.8409	-2.0970	-2.4690
4	-0.700429	$-0.308\ 815$	-0.388776	-0.309050	-0.389019	-2.8374	-2.3594	-2.5984
5	$-0.713\ 143$	-0.316 049	-0.394444	-0.316 183	-0.394530	-2.6498	-2.4312	-2.5405
4-5	-0.726483	-0.323639	-0.400391	-0.323666	$-0.400\ 311$	-2.4529	-2.5066	-2.4798
3-4	$-0.727\ 609$	-0.323654	$-0.401\ 121$	-0.323780	-0.401279	-2.8348	-2.5508	-2.6928
2-3	-0.711383	-0.313057	-0.395448	-0.313358	-0.395649	-2.8775	-2.3751	-2.6263
				$(H_2O)_2$				
2 ^a	-0.466934	$-0.232\ 313$	$-0.232\ 313$	-0.232676	$-0.233\ 031$	-2.3083	-1.2262	-1.7672
3 ^a	$-0.565\ 882$	-0.281676	-0.281676	-0.281949	$-0.282\ 041$	-2.5295	-1.8919	-2.2107
4 ^a	-0.597550	-0.297581	-0.297581	-0.297~675	-0.297754	-2.3881	-2.1213	-2.2547
3-4	$-0.620\ 660$	$-0.309\ 187$	$-0.309\ 187$	$-0.309\ 150$	$-0.309\ 221$	-2.2849	-2.2887	-2.2868
2-3	-0.607544	-0.302 461	-0.302 461	-0.302 677	-0.302 695	-2.6226	-2.1722	-2.3974

^aFrom Ref. 17.

correction, the X^{-3} form describes the basis set convergence of the interaction energy very well.

Comparing the extrapolated results with the averages of $\Delta E_{\rm corr}$ and $\Delta E_{\rm corr}^{\rm CP}$, we see that at the aug-cc-pVTZ level, the simple average performs better than the 2–3 extrapolated $\Delta E_{\rm corr}^{\rm CP}$ results, which is not that surprising since the extrapolation is known to degrade in performance when double-zeta results are included.^{45,47} Otherwise, the simple average at the aug-cc-pVQZ level and the 3–4 extrapolated $\Delta E_{\rm corr}^{\rm CP}$ results are very similar, and also the simple average at the aug-cc-pV5Z level and the 4–5 extrapolated $\Delta E_{\rm corr}^{\rm CP}$ results are very close to each other.

D. Convergence and extrapolations of the CCSD(T) correlation contribution

The results for the CCSD(T) correlation contributions are given in Table V and illustrated in Figs. 11–15. The situation for the CCSD(T) correlation contributions is prac-

tically the same as the one for the MP2 correlation contributions, which is not unexpected as the CCSD(T) and MP2 correlation energies usually display similar basis set converge.⁴⁵ Again, the convergence of ΔE_{corr}^{CP} is always monotonic, smooth, slow, and from above, with large errors at the aug-cc-pVDZ basis set, and the convergence of ΔE_{corr} is always nonmonotonic, decreasing from aug-cc-pVDZ to aug-cc-pVTZ and increasing from aug-cc-pVTZ to aug-ccpV5Z. The performance of the extrapolations is therefore also virtually the same as at the MP2 level, and the 4-5 extrapolated ΔE_{corr}^{CP} results are therefore used in Figs. 11–14 as representatives of the basis set limit (the horizontal dashed lines), which, according to the findings above, should be accurate to at least 0.05 kcal/mol. Due to the unsystematic convergence of $\Delta E_{\rm corr}$, the two-point X^{-3} extrapolation based on ΔE_{corr} results does not perform very well, whereas the extrapolation based on $\Delta E_{\text{corr}}^{\text{CP}}$ results and the simple av-erage of ΔE_{corr} and $\Delta E_{\text{corr}}^{\text{CP}}$ both perform very well. Only in a



FIG. 11. $\Delta E_{\rm corr}^{\rm CP}$ (upper curve) and $\Delta E_{\rm corr}$ (lower curve) for (HF)₂ at the CCSD(T) level as a function of the cardinal number X. The unconnected diamonds (\diamond) are the simple average of the $\Delta E_{\rm corr}^{\rm CP}$ and $\Delta E_{\rm corr}$ results, the unconnected crosses (×) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, the asterisks (*) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm cP}$ results, and the horizontal dashed line is the 4–5 $\Delta E_{\rm corr}^{\rm CP} X^{-3}$ extrapolated result.

few cases (at the aug-cc-pVTZ level) the simple average performs noteworthy better than the extrapolation based on $\Delta E_{\rm corr}^{\rm CP}$ results; otherwise, these two schemes give results that are very close to each other. Finally, at the aug-cc-pV5Z level, the simple average is always slightly lower than the 4–5 $\Delta E_{\rm corr}^{\rm CP}$ extrapolated result, which is also in accordance with our findings at the MP2 level.

For the correlation contribution to the interaction energy, the basis set errors are much larger than for the SCF part. The reduction of the basis set errors for the correlation contribution is thus very important. For the aug-cc-pVDZ basis set, the basis set errors of the ΔE_{corr} results are significantly smaller than those of the ΔE_{corr}^{CP} results, and the ΔE_{corr} results are also better than the simple average of ΔE_{corr} and ΔE_{corr}^{CP} . The good performance of ΔE_{corr} at the aug-cc-pVDZ level, however, is due to a fortuitous cancellation of errors between BSSE and the other (ordinary) basis set errors, and the augcc-pVDZ basis set is in general not sufficiently reliable for high-accuracy studies (see the results for HCNHF in Table



FIG. 12. $\Delta E_{\rm corr}^{\rm CP}$ (upper curve) and $\Delta E_{\rm corr}$ (lower curve) for (HCl)₂ at the CCSD(T) level as a function of the cardinal number X. The unconnected diamonds (\diamond) are the simple average of the $\Delta E_{\rm corr}^{\rm CP}$ and $\Delta E_{\rm corr}$ results, the unconnected crosses (\times) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, the asterisks (*) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm cpr}$ results, and the horizontal dashed line is the 4–5 $\Delta E_{\rm corr}^{\rm CP} X^{-3}$ extrapolated result.



FIG. 13. $\Delta E_{\rm corr}^{\rm CP}$ (upper curve) and $\Delta E_{\rm corr}$ (lower curve) for H₂OHF at the CCSD(T) level as a function of the cardinal number X. The unconnected diamonds (\diamond) are the simple average of the $\Delta E_{\rm corr}^{\rm CP}$ and $\Delta E_{\rm corr}$ results, the unconnected crosses (\times) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, the asterisks (*) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, and the horizontal dashed line is the 4–5 $\Delta E_{\rm corr}^{\rm CP}$ X^{-3} extrapolated result.

III, and Figs. 9 and 14, where the basis set error of the aug-cc-pVDZ ΔE_{corr} result is about 0.2 mE_h). For larger systems whose size prohibits the use of basis sets larger than the aug-cc-pVDZ set, the ΔE_{corr} results obtained in the aug-cc-pVDZ basis set are expected to be the ones with the smallest basis set error. However, for systems where the larger basis sets can be employed, both the simple average of ΔE_{corr} and ΔE_{corr}^{CP} and the two-point X^{-3} extrapolated ΔE_{corr}^{CP} results have significantly smaller basis set errors than the original ΔE_{corr} and ΔE_{corr}^{CP} results. At the aug-cc-pVTZ level, both schemes give results that are within 0.2 kcal/mol of the basis set limit. Both schemes are thus very effective and useful in reducing the basis set error for the correlation contribution to the interaction energy.



FIG. 14. $\Delta E_{\rm corr}^{\rm CP}$ (upper curve) and $\Delta E_{\rm corr}$ (lower curve) for HCNHF at the CCSD(T) level as a function of the cardinal number X. The unconnected diamonds (\diamond) are the simple average of the $\Delta E_{\rm corr}^{\rm CP}$ and $\Delta E_{\rm corr}$ results, the unconnected crosses (×) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, the asterisks (*) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, and the horizontal dashed line is the 4–5 $\Delta E_{\rm corr}^{\rm CP}$ X^{-3} extrapolated result.



FIG. 15. $\Delta E_{\rm corr}^{\rm CP}$ (upper curve) and $\Delta E_{\rm corr}$ (lower curve) for (H₂O)₂ at the CCSD(T) level as a function of the cardinal number X. The unconnected diamonds (\diamond) are the simple average of the $\Delta E_{\rm corr}^{\rm CP}$ and $\Delta E_{\rm corr}$ results, the unconnected crosses (×) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, the asterisks (*) are the two-point X^{-3} extrapolated results based on the $\Delta E_{\rm corr}^{\rm CP}$ results, and the horizontal dashed line is the 3–4 $\Delta E_{\rm corr}^{\rm CP} X^{-3}$ extrapolated result.

IV. CONCLUSION

The basis set convergence of the SCF and correlation contributions to the interaction energy of the five hydrogenbonded complexes $(HF)_2$, $(HCl)_2$, H_2OHF , HCNHF, and $(H_2O)_2$ has been investigated via conventional SCF, MP2, and CCSD(T) calculations for the aug-cc-pVXZ series of basis sets and explicitly correlated MP2-R12/A calculations. At the SCF level, the CP and uncorrected interaction energies both converge very unsystematically. The basis set errors in the SCF part, however, are significantly smaller than those in the correlation contribution, and the latter errors are therefore the most important. For the correlation contribution, the CP and uncorrected results converge toward the basis set limit from each side (the former from above), and their simple average is therefore remarkably close to the basis set limit.

The basis set convergence of the CP correlation contribution is very different from that of the uncorrected correlation contribution. While the latter is always nonmonotonic and very unsystematic, the former is always monotonic, smooth, slow, and systematic. The unsystematic convergence of the uncorrected results is a consequence of the combination of two different basis set errors; BSSE and the errors associated with the incomplete description of the electronic Coulomb cusp. However, once BSSE has been effectively removed by the counterpoise correction, the cusp dominates and the convergence becomes systematic and completely like that of the correlation energy. Consequently, the basis set convergence of the CP correlation contribution follows an X^{-3} form similar to the one for the correlation energy, and two-point extrapolations based on this form yield results that are closer to the basis set limit than the original CP and uncorrected correlation contributions are. In particular, when double-zeta results are not included in the extrapolations, results within 0.05 kcal/mol of the basis set limit are obtained.

Although the convergence of the uncorrected correlation contribution is very unsystematic, those results are never more than 0.3 kcal/mol from the basis set limit, while the CP results within the aug-cc-pVDZ basis set have errors larger than 0.5 kcal/mol. For the smaller basis sets, the uncorrected results are closer to the basis set limit than are the CP results, but this is due to a fortuitous cancellation of errors. Following the discussion above, the basis set convergence of the CP correlation contribution is dominated by the incomplete description of the electronic Coulomb cusp, and the error associated with the cusp makes the correlation contribution to the interaction energy too high-see, e.g., Figs. 6-15. On the other hand, BSSE makes the interaction energy too low (compare the CP and uncorrected correlation contributions). The cancellation of these two oppositely directed errors leads to a fortuitously good performance of the uncorrected interaction energies-as well as the very unsystematic convergence of these-and the uncorrected results in the small basis sets should therefore be judged and employed with caution.

Finally, our most accurate results for the equilibrium interaction energies of the five complexes are obtained from combination of the SCF $\Delta E^{\text{ave}}(5)$ results and the 4–5 $\Delta E^{\text{CP}}_{\text{corr}}$ extrapolated results at the valence only CCSD(T) level (3–4 extrapolated results for (H₂O)₂]. These results are: -4.53 for (HF)₂, -1.95 for (HCl)₂, -8.51 for H₂OHF, -7.31 for HCNHF, and -4.96 kcal/mol for (H₂O)₂.

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