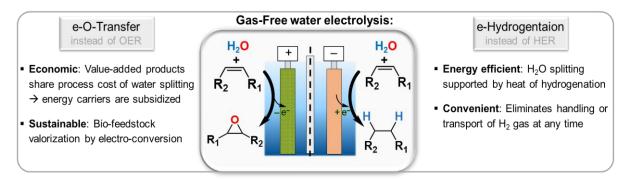
Transition Metal Catalyzed Electro-Organic Synthesis with Water & Ex Situ Electrosynthesis for Unique Selectivity

Dr. Dominik P. Halter, Technical University of Munich, Garching (Germany).

TUM School of Natural Sciences, Department of Chemistry & Catalysis Research Center (CRC) Lichtenbergstr. 4, 85748 E-Mail: <u>dominik.halter@tum.de</u>

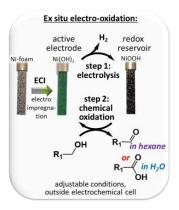


 H_2 production by water electrolysis is proposed to store excess renewable electricity in chemical energy carriers. O_2 evolution is the typically paired anode reaction in standard electrolyzers, but the unsalable side product O_2 cannot support process costs.

We develop electrocatalytic oxygen atom transfer (e-OAT), to replace O_2 evolution by anodic valueadded product formation together with water electrolysis-based energy carrier production. A specific target reaction is the electro-epoxidation of alkenes with H_2O .^[1]

Besides cost, H_2 gas storage is another challenge, for which liquid organic hydrogen carriers (LOHCs) emerged as a potent solution. In this context, we develop **direct electro-hydrogenation of alkenes**,^[2] **organic carbonyls, and aromatic systems with H_2O in one step**, mediated by transition metal pincer complexes and porphyrins. Mild overpotentials (energy efficiency) and suppression of typically dominant parasitic HER (current efficiency), is achieved by structure-reactivity tuning of catalysts and the design of unique mechanisms, including H_2O activation instead of H^+ reduction.

Beyond classic *in situ* electro-synthesis (substrates added into electrolyzer), we develop *ex situ* electrosynthesis—a new concept to separate chemical and electrochemical reaction steps in time and space for unrestricted flexibility in reaction design.^[3] Exemplarily, we use the reversible Ni(OH)₂ / NiOOH couple for short term storage of redox equivalents. The bulk electro-conversion of Ni(OH)₂ to NiOOH is performed together with energy carrier (H₂) production in an electrolyzer. The chemical oxidation of organic substrates to value-added products is then performed outside the electrolyzer under freely choosable reaction conditions (solvent, temperature, pressure), which allows orthogonal selectivity for products that are inaccessible to traditional electrosynthesis.



References:

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