Synthesis of a well-defined polyelectrolyte by controlled/ "living" nitroxide mediated radical polymerization. Kinetic study

 Christos Pantazidis^a, Stelios Andreou^a, Emmanouil Glynos^b, Georgios Sakellariou^a
^a Department of Chemistry, National and Kapodistrian University of Athens, Panepistimiopolis Zografou, 15771 Athens, Greece
^b Institute of Electronic Structure and Laser, Foundation for Research and Technology – Hellas, P.O. Box 1385, Heraklion, Crete 71110, Greece

The synthesis and characterization 4-styrenesulfonyl of potassium (trifluoromethylsulfonyl) imide (STFSIK) monomer as well as its controlled/living nitroxide-mediated radical polymerization (NMRP), using N-tert-Butyl-N-(2-methyl-1-phenylpropyl)-O-(1-phenylethyl) hydroxylamine as initiator in 50% w/v solutions of DMF at 125 °C, are reported. The kinetic studies of STFSIK indicated controlled/living polymerization process free of coupling and transfer reactions. ¹H-NMR and SEC were employed to determine the rate of monomer consumption and molecular weight distribution, D, of the synthesized samples. The polymerization reaction established high monomer conversion in just an hour with high end-group fidelity and the ability to continue polymerization with no visible terminations. Finally, the synthesis of the corresponding PSTFSILi homopolymer was achieved through a metathesis reaction between PSTFSIK and LiClO₄, and the subsequent purification was successful against dialysis in water.

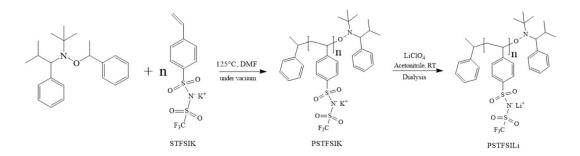


Figure 1: Polymerization of STFSIK and subsequent ion exchange to produce PSTFSILi

References:

 C. Pantazidis, S. Andreou, E. Glynos, G. Sakellariou, European Polymer Journal 134,109815 (2020)